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Updated: 01/12/2015

# COVALENT COMPOUND HAIKU

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## ● **PRESENTATION1**

Compounds of TEST1 What is a Compound- Chemical formulas subscripts coefficients  
2 Covalent compound- What are they made of- How do they form- Naming Formula writing  
3 Polyatomic Ions- - What are they made of- Naming formula writing  
4 Ionic compound- What are they made of- How do they form- Naming Formula writing  
5 Acids- What are they made of- naming formula writing  
Covalent Compounds Also called Molecular...

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## ● **11121/13**

CHEM 4 Quiz 8 (20 points) Name IJ1S WV ke y ~~~~~r----- 1. (5 points) Provide a term or a name that matches each of the word descriptions below. An electrically neutral group of atoms held together by covalent bonds acting as the smallest independent particle of a chemical substance. A chemical compound that has no ions in it. A chemical substance that is made of atoms with the sa...

### **File name:** Quiz8\_Chem4\_F2013\_Ans\_Key.pdf

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## ● **UNTITLED**

In a periodic table square, what number tells you how many grams equals one Atomic Mass mole? 1 mole C = ??? grams 12  
1 mole K = ??? grams 39 1 mole He = ???grams 4 2 moles Cl = ??? grams 70 When referring to a gas, STP means....  
Standard Temperature & Pressure Standard temperature is... 0 degrees Celsius or 273 Kelvins Standard pressure is .... 1 atm  
or 101.3 kPa or 760 mmHg One unit or piece of ...

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- **MICROSOFT WORD - PSJAN05SOLN.DOC**

January 2005430 Solutions1. D2. A3. D4. B5. D6. C7. C and H are both non-metals. They form a covalent compound with carbon forming an octet of electrons. Hydrogen seeks to fill its energy level with two electrons. H C C H8. Cl<sub>2</sub> = 35.5 g mole Cl \* 2mole Cl mole Cl<sub>2</sub> = 71 g mole Cl<sub>2</sub> ( mole Cl<sub>2</sub> 6.02 X10<sup>23</sup> molecules) = 71 6.02 X10<sup>23</sup> g molecule = 1.18 X 10<sup>-22</sup> g molecule9. Li is bigger than...

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- **NAME \_\_\_\_\_ AGENT NAME \_\_\_\_\_**

Your Mission Creating Covalent BondsUse your covalent bonding cards to figure out how atoms will form covalent molecules Record informationabout the each of the compounds you create1 A covalent compound formed between Nitrogen and ChlorineLewis Dot Draw the lewis dot formula for your Draw the structural formula and thestructures molecule below Circle shared pairs chemical formula for this molecule...

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- **CLIN CHEM 28 9 1844-1847 1982**

Analytical ConceptsRelease Tags A New Classof AnalyticalReagentsRegula Joppich-Kuhn Markus Joppich and Roger W GleseRelease tags are novel compounds made up of three SIGNAL L JEASE L J REACTIVITYmolecular groups signal release and reactivity GROUPJf GROUP IGROUPTThe signal group is the detected part of the tag the releaseRELEASE TAG Sgroup provides a site for specific covalent cleavage andthe react...

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