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Honour Chemistry: Unit 4 Practice Test: Thermochemistry and Nuclear ChemistryPart A: Multiple Choice and Numerical Response (1 mark each for Multiple Choice) (2 marks each for Numerical Response) $E_{\text{bind}} = mc^2$ $c = \text{speed of light (3.00 } 10^8 \text{ m s}^{-1})$ $1 \text{ eV} = 1.69 \cdot 10^{-19} \text{ J}$ ($1 \text{ kg} = 6.022 \cdot 10^{26} \text{ amu}$, or $1 \text{ g} = 6.022 \cdot 10^{23} \text{ amu} = 1 \text{ mole amu}$) $1 \text{ MeV} = 1.69 \cdot 10^{-16} \text{ J}$ ($1 \text{ Neutron} = 1.008665 \text{ amu}$) ($1 \text{ Proton} = 1.007825 \text{ amu}$...

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- **[MICROSOFT WORD - GRADE 10 CHEMISTRY UNIT REVIEW.DOC](#)**

Test Date: Friday, November 21, 2008 Name: Grade 10 Academic Science (SNC 2D1) CHEMISTRY UNIT REVIEW This review is simply a guide to the information that we have covered. It is by no means a repeat of the test to come. Make sure that you study all the material that we have learned this unit, using this only as a supplementary resource. Start studying now! 1. There are different types of compounds...

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- **[UNIT 4 THERMOCHEMISTRY AND NUCLEAR CHEMISTRY NOTES \(ANSWERS\)](#)**

Chemistry AP Unit 4: Thermochemistry and Nuclear Chemistry Unit 4: THERMOCHEMISTRY AND NUCLEAR CHEMISTRY Chapter 6: Thermochemistry 6.1: The Nature of Energy Energy (E): - the ability to do work or produce heat. First Law of Thermodynamics: - states that energy cannot be created or destroyed. It can only be converted from one form to another. Therefore, energy in the universe is a constant. - also kno...

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- **[UNIT 1: BASIC CHEMISTRY NOTES \(ANSWERS\)](#)**

Honour Chemistry Unit 1: Basic Chemistry Unit 1: BASIC CHEMISTRY Chapter 3: Scientific Measurement 3.1: The Importance of Measurement Qualitative Measurements: - measurements that do not involve a numerical value. (Examples: Colour, Odour, Heat Given off or Taken in, Type of Solid Formed etc) Quantitative Measurements: - measurements that do include a numerical value. (Examples: Volume, Temperature, M...

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